 Cambridge Assessment International Education

Cambridge IGCSE™

CO-ORDINATED SCIENCES **0654/62**

Paper 6 Alternative to Practical **October/November 2022**

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2022 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

This document consists of **10** printed pages.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1: Marks must be awarded in line with: • the specific content of the mark scheme or the generic level descriptors for the question • the specific skills defined in the mark scheme or in the generic level descriptors for the question • the standard of response required by a candidate as exemplified by the standardisation scripts.
GENERIC MARKING PRINCIPLE 2: Marks awarded are always **whole marks** (not half marks, or other fractions).
GENERIC MARKING PRINCIPLE 3: Marks must be awarded **positively**: • marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate • marks are awarded when candidates clearly demonstrate what they know and can do • marks are not deducted for errors • marks are not deducted for omissions • answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.
GENERIC MARKING PRINCIPLE 4: Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	'List rule' guidance For questions that require *n* responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards *n*. • Incorrect responses should not be awarded credit but will still count towards *n*. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

6	Calculation specific guidance
	Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.
	For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.
	For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.
	Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.
7	Guidance for chemical equations
	Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.
	State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Question	Answer	Marks
1(a)(i)	left hand side ;	**1**
1(a)(ii)	6.8 ;	**1**
1(a)(iii)	the size of coloured liquid / bubble will affect the reading / can compare results ;	**1**
1(b)(i)	4.4 ;	**1**
1(b)(ii)	0.44 ;	**1**
1(c)	readings less / smaller distance/number ;	**1**
1(d)	volume decreases / pressure decreases / **measuring amount** of oxygen used up ;	**1**
1(e)	to allow in oxygen / animals can breathe / do not die / suffocate ;	**1**

Question	Answer	Marks
2(a)(i)	metre rule / measuring tape ;	**1**
2(a)(ii)	100 ;	**1**
2(b)(i)	12 ;	**1**
2(b)(ii)	9 ;	**1**
2(c)	900 ;	**1**

Question	Answer	Marks
3	**1 point from each section plus any other 2 points** **apparatus** measuring cylinder ; stop-watch/stop-clock/timer ; **method** mixing starch solution, enzyme, and iodine **AND** start stop clock ; repeats at same enzyme volume / concentration ; more than one volume / concentration of amylase ; goggles so that enzyme does not get into eye ; **measurements** time for disappearance of blue-black colour / for colour to go brown ; volume of amylase **AND** volume of starch solutions ; **control variables** volume of starch solution / concentration of starch ; volume of enzyme ; temperature ; pH ; **processing and conclusion** identify/exclude anomalies and calculate average ; draw graph of concentration/volume against time ; as concentration of enzyme increases/decreases look for the pattern in rate/time ; **OR** *alternative method* **apparatus** spotting tile / method of sampling such as a dropping pipette ; measuring cylinder ; stop-watch/stop-clock/timer ;	**7**

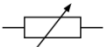| Question	Answer	Marks
3	**method** mixing starch solution with enzyme **AND** start stop clock ; taking regular samples e.g. per minute and tests with iodine ; repeats at same enzyme volume / concentration ; more than one volume or concentration of amylase ; **measurements** time for blue-black colour not to appear / for iodine to (just) stay brown ; volume of amylase and volume of starch solutions ; **control variables** volume / concentration of starch solution ; volume of enzyme ; temperature ; pH ; **conclusion** calculate average to identify/exclude anomalies ; draw graph of concentration / volume against time ; as concentration of enzyme increases/decreases look for the pattern in rate / time ;	

Question	Answer	Marks
4(a)(i)	tripod, gauze and Bunsen↑ ; evaporating basin and copper carbonate in it ; 2 correct labels ;	**3**
4(a)(ii)	so **all** powder reacts / ensure **complete** reaction ;	**1**
4(a)(iii)	25.91 ; 25.23 ;	**2**

Question	Answer	Marks
4(a)(iv)	gas/CO_2 escapes/released (into the air) / gas/CO_2 lost ;	1
4(a)(v)	all black / no green ;	1
4(b)(i)	2.45 ;	1
4(b)(ii)	1.568 / 1.57 / 1.6 ;	1
4(b)(iii)	1.77 ;	1
4(b)(iv)	112.8188776 % ; 113 % ;	2
4(c)	not all copper carbonate decomposed / impure copper carbonate ;	1
4(d)	repeat (and average) ;	1

Question	Answer	Marks
5(a)	(nitric acid and) silver nitrate **AND** white ppt ; others have cream (ppt) and colourless (solution) ;	2
5(b)	(nitric acid and) barium nitrate **AND** white ppt ; others give colourless (solution) / no reaction ; **OR** (nitric acid) and silver nitrate **AND** cream ppt ; others have a white ppt and colourless solution ;	2
5(c)	flame is almost the same colour (as Bunsen flame) / colour of (Bunsen) flame masks the colour ;	1

Question	Answer	Marks
6(a)(i)	1.3 ;	1
6(a)(ii)	4.<u>0</u> ;	1
6(a)(iii)	image is bigger / magnified / image is inverted ;	1
6(b)	3.1 ;	1
6(c)(i)	axes have the correct orientation and are labelled with a quantity and unit ; suitable linear scales, where points cover $\geqslant$ ½ the grid used ; 5 points plotted correctly to $\pm$ ½ small square ;	3
6(c)(ii)	smooth curve of best fit ;	1
6(d)(i)	u = 45 $\pm$ 1 (cm) ;	1
6(d)(ii)	15(.0) (cm) ;	1
6(e)	hand / ruler is in the way of the image / image is small / dim ;	1

Question	Answer	Marks
7(a)(i)	correct parallel connection with correct voltmeter symbol ;	1
7(a)(ii)	1.6 ; 0.96 ;	2
7(b)	1.7 Ω ;	1
7(c)	5.0 $\pm$ 0.4 ;	1
7(d)	agree – doubling l (approximately) doubles R (or vice versa) / ratio R / l is (approximately) constant (or vice versa) ;	1

Question	Answer	Marks
7(e)	difficult to judge position of the crocodile clip / difficult to measure to nearest mm ;	1
7(f)	correct symbol for variable resistor ; in correct series connection with the rest of the circuit ;	2